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(54) Title: METHOD FOR MANUFACTURE OF SERTINDOLE

(57) Abstract: The present invention relates to a novel method for manufacture of sertindole comprising manufacturing 5-chloro-1-(4-fluorophenyl)-indole and converting it to sertindole wherein the method for manufacture of 5-chloro-1-(4-fluorophenyl)-indole comprises reacting 5-chloro-indole with a 4-fluorophenylhalide in the presence of a base, a chelating ligand and catalytic amounts of a copper salt comprising copper(I) or copper(II) and an anion which does not interfere in an unfavourable way with the reaction.



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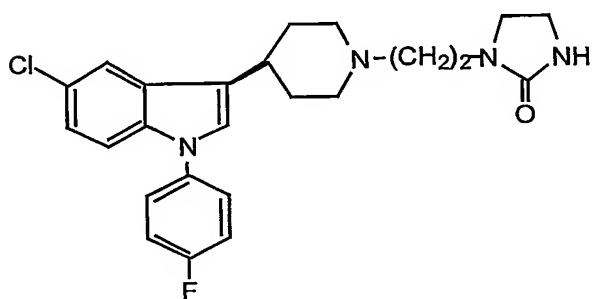
Method for manufacture of sertindole

Field of the invention

The present invention relates to a new method of manufacturing the compound
1-[2-[4-[5-chloro-1-(4-fluorophenyl)-1-*H*-indol-3-yl]-1-piperidiny]ethyl]-2-imidazolidinone
having the INN name sertindole and a new method of manufacturing the intermediate, 5-
chloro-1-(4-fluorophenyl)-indole used in the method.

Background of the invention

Sertindole is a well-known antipsychotic drug having the formula



The compound was disclosed in US patent No 4,710,500 and the antipsychotic activity thereof was described in US patent No 5,112,838. Sertindole is a potent centrally acting 5-HT₂ receptor antagonist in vivo and has further been disclosed to be active in models indicative of effects in the treatment of anxiety, hypertension, drug abuse and cognitive disorders.

A number of syntheses of sertindole have been disclosed in US patent No 4,710,500 and WO 98/51685. 5-chloro-1-(4-fluorophenyl)-indole is a key intermediate in these syntheses. The syntheses of 5-chloro-1-(4-fluorophenyl)-indole as disclosed in US patent No 4,710,500 and WO 98/51685 require multiple steps from commercially available starting materials, are expensive, occupy production equipment for prolonged periods resulting in low production capacity and result in environmental impact and safety. The synthesis which has been favoured so far for industrial synthesis of sertindole comprises the multiple step synthesis of 5-chloro-1-(4-fluorophenyl)-indole as disclosed in WO 98/51685.

An alternative synthetic strategy for 1-aryl-indoles is the Ullmann arylation of N-unsubstituted indoles with aryl halides catalyzed by large amounts of copper, typically near-stoichiometric amounts or more, as disclosed in e.g. *J. Med. Chem.* 1992, 35 (6), 1092-1101. The Ullmann arylation has, however, hitherto been disfavoured with regards to the synthesis of 5-chloro-1-(4-fluorophenyl)-indole due to various problems which to those skilled in the art are known to apply to the Ullmann arylation in general as the reactions typically result in

moderate yields, around 50%, correspondingly large amounts of coloured by-products and cumbersome work-up procedures caused by the complexation of the reaction product with the copper catalyst. These complexes often require surprisingly harsh treatment to liberate the free reaction product, as known to those skilled in the art.

Hence, there is a desire for new methods for manufacturing of 5-chloro-1-(4-fluorophenyl)-indole. Such new methods may be advantageous in that they are more cost effective, require fewer reaction steps, have reduced impact on the environment, give higher yields, result in increased production capacity, purer crude product and easier work-up procedures.

Recently, Klapars et al. *J.Am.Chem.Soc.* 2001, 123, 7727-7729, disclosed a variant of the Ullmann arylation wherein copper is present in catalytic amounts together with the chelating ligand trans-1,2-cyclohexanediamine.

Summary of the invention

It has now surprisingly been found that it is possible to manufacture 5-chloro-1-(4-fluorophenyl)-indole in an efficient way giving good yields by arylation of 5-chloro-indole with a 4-fluorophenylhalide in the presence of catalytic amounts of a copper salt and a chelating ligand. This reaction is surprisingly selective. Illustrative of this high selectivity is the fact that there is virtually no by-products formed by reaction between the 5-chloro group of one molecule of 5-chloro-indole and the nitrogen of another molecule of 5-chloro-indole. This type of side reaction would be expected from the disclosure in *J.Am.Chem.Soc.* 2001, 123, 7727-7729, which illustrate the reactivity of arylchlorides in this type of reactions. It has even more surprisingly been found that the chelating ligand may be as simple as ethylenediamine. This reaction gives 5-chloro-1-(4-fluorophenyl)-indole in high yields and purity in a cost-effective single-step synthesis from commercially available starting materials.

Hence, the present invention relates to a novel method for manufacture of sertindole comprising manufacturing 5-chloro-1-(4-fluorophenyl)-indole and converting it to sertindole wherein the method for manufacture of 5-chloro-1-(4-fluorophenyl)-indole comprises reacting 5-chloro-indole with a 4-fluorophenylhalide in the presence of a base, a chelating ligand and catalytic amounts of a copper salt comprising copper(I) or copper(II) and an anion which does not interfere in an unfavourable way with the reaction.

Furthermore, the present invention relates to a method for manufacture of 5-chloro-1-(4-fluorophenyl)-indole comprising reacting 5-chloro-indole with a 4-fluorophenylhalide in the presence of a base, a chelating ligand and catalytic amounts of a copper salt comprising copper(I) or copper(II) and an anion which does not interfere in an unfavourable way with the reaction.

Detailed description of the invention

As used throughout the description and the claims, the following definitions apply:

The term '4-fluorophenylhalide' means any compound selected from the group consisting of 4-fluoro-chlorobenzene, 4-fluoro-bromobenzene and 4-fluoro-iodobenzene.

5 The term 'catalytic amounts' means amounts that are significantly lower than stoichiometric amounts such as less than 20 mol % relative to 5-chloro-indole.

The term 'chelating ligand' means any compound comprising at least two atoms that are able to simultaneously coordinate to the same metal atom.

10 The term 'C₁₋₆-alkyl' refers to a branched or unbranched alkyl group having from one to six carbon atoms inclusive, such as methyl, ethyl, 1-propyl, 2-propyl, 1-butyl, 2-butyl, 2-methyl-2-propyl, and 2-methyl-1-propyl.

The term 'C₁₋₆-alkyl carboxylic acid' refers to C₁₋₆-alkyl groups which are terminated by a carboxylic acid.

15 The term 'aryl' refers to a carbocyclic aromatic group, such as phenyl or naphthyl, in particular phenyl.

In one aspect the present invention relates to a method for manufacture of sertindole comprising manufacturing 5-chloro-1-(4-fluorophenyl)-indole and converting it to sertindole wherein the method for manufacture of 5-chloro-1-(4-fluorophenyl)-indole comprises reacting 5-chloro-indole with a 4-fluorophenylhalide in the presence of a base, a chelating
20 ligand and catalytic amounts of a copper salt comprising copper(I) or copper(II) and an anion which does not interfere in an unfavourable way with the reaction.

In a further aspect, the present invention relates to a method for manufacture of 5-chloro-1-(4-fluorophenyl)-indole comprising reacting 5-chloro-indole with a 4-fluorophenylhalide in the presence of a base, a chelating ligand and catalytic amounts of a
25 copper salt comprising copper(I) or copper(II) and an anion which does not interfere in an unfavourable way with the reaction.

The embodiments described hereafter applies to all aspects of the invention.

In one embodiment of the invention, the chelating ligand is a substituted or unsubstituted 1,10-phenanthroline, such as an unsubstituted 1,10-phenanthroline. In another
30 embodiment the chelating ligand is a compound of the formula $X-(CR^1R^2-(CR^5R^6)_n-CR^3R^4-Y)_m$, wherein X and Y independently are selected from NR^7R^8 and OR^9 , R^1-R^9 independently are selected from hydrogen, C₁₋₆-alkyl, C₁₋₆-alkyl carboxylic acid and aryl or one of R^1 and R^2 together with one of R^5 and R^6 are C₃₋₆-alkylene, m is 1 or 2, and n is 0, 1, 2 or 3. In a preferred embodiment, at least one of X and Y is NR^7R^8 , and more preferred both of X and Y
35 are NR^7R^8 . In another preferred embodiment, R^7 and R^8 are independently selected from hydrogen, C₁₋₆-alkyl and C₁₋₆-alkyl carboxylic acid, and more preferred R^7 and R^8 are

hydrogen. In yet another preferred embodiment, R⁵ and R⁶ are hydrogen. In yet another preferred embodiment, m is 1. In yet another preferred embodiment, n is 0. In yet another preferred embodiment R¹-R⁴ are hydrogen, or R¹ and R³ together are C₃₋₆-alkylene and R² and R⁴ are hydrogen. Preferred chelating ligands are those selected from the group comprising
5 1,2-cyclohexanediamine, N,N,N,N-tetramethyl ethylenediamine, N,N-diethyl ethylenediamine, ethylenediamine, ethylenediamine N,N,N,N-tetraacetic acid (EDTA), diethylenetriamine N,N,N,N,N-pentaacetic acid (DTPA) and substituted or unsubstituted 1,10-phenantroline; more preferred chelating ligands are those selected from the group comprising 1,2-cyclohexanediamine, N,N,N,N-tetramethyl ethylenediamine, N,N-diethyl
10 ethylenediamine and ethylenediamine, and the most preferred chelating ligand is ethylenediamine.

In a preferred embodiment of the invention, the 4-fluorophenylhalide is 4-fluorobromobenzene or 4-fluoro-iodobenzene as the reactivity of the 4-fluorophenylhalides increases in the order chloro-<bromo-<iodo for this type of reactions. In a preferred
15 embodiment of the invention the 4-fluorophenylhalide is added in a molar surplus relative to 5-chloro-indole. Preferably the molar ratio 4-fluorophenylhalide:5-chloro-indole is in the range from about 1.1 to about 3, more preferred from about 1.2 to about 2.5, and most preferred from about 1.3 to about 2.0.

The methods of manufacture according to the present invention are advantageous as
20 compared to classical Ullmann arylation as they only require catalytic amounts of a copper salt, i.e. less than 20 mol % relative to 5-chloro-indole. Preferably the amount of copper salt is less than 10 mol % relative to 5-chloro-indole and even more preferred in the range from about 1 to about 5 mol %. The products made according to the present invention may be isolated without the harsh treatment, such as boiling in hydrochloric acid or treatment with
25 cyanides, which often is necessary in order to break the complexes between copper and the product of the classical Ullmann reactions.

Any copper salt comprising copper(I) or copper (II) and an anion which does not interfere in an unfavourable way with the reaction may be applied. Exemplary of anions, which may interfere in an unfavourable way with the reaction, are cyanide, sulphide and
30 selenide. Cyanide may react as a nucleophile and compete with the indole for reaction with the 4-fluorophenylhalide, whereas sulphide and selenide may inactivate the copper catalyst. Those skilled in the art will be aware that other anions also may interfere in an unfavourable way with the reaction and easily realise if an anion interferes in an unfavourable way with the reaction. Preferred copper salts for use in the present invention are selected from the group
35 comprising CuCl, CuBr, CuI, CuCl₂, CuBr₂, CuI₂, CuOCOCH₃, Cu(OCOCH₃)₂, anhydrous or hydrated CuSO₄, CuCO₃, Cu₂O and mixtures of said copper salts; more preferred copper salts

are those selected from the group comprising CuCl, CuBr, CuI, CuCl₂, CuBr₂ and CuI₂. These work well as catalysts in the reaction and are readily available to reasonable prices. The copper salt may be added in one portion at the start of the reaction or in two or more portions distributed over the reaction time.

5 Various bases may be employed in the methods of manufacture of the present invention. Exemplary bases are the carbonates, hydrogen carbonates, phosphates, hydrogen phosphates, dihydrogen phosphates, oxides and hydroxides of alkali metals. Preferred bases are potassium and sodium carbonates as these are readily available to a low price and easy to handle. The base is typically present in a molar excess relative to 5-chloro-indole, preferably
10 the amount of base is in the range from about 1.05 molar equivalents to about 2.5 molar equivalents.

The methods of manufacture of the present invention may be performed by heating a neat mixture of the reactants without any solvent or in a suitable solvent system. Exemplary
15 of such solvent systems are toluene, mixtures of toluene and water, ethers such as dioxane, tetrahydrofuran (THF), diethyl ether, dimethyl ether, monoethylene glycol dimethyl ether (monoglyme) and diethylene glycol dimethyl ether (diglyme), amides such as dimethylformamide (DMF), dimethylacetamide (DMA), N-methyl-pyrrolidone (NMP). Preferred solvents are DMF and toluene and most preferred is DMF.

Typically the methods of manufacture of the present invention are performed at
20 temperatures above 80 °C, preferably in the range from 90 °C to 200 °C, more preferred in the range from 100 °C to 160 °C. Higher yields may be obtained by pretreating the reaction system at a temperature in the range from about 30 °C to about 70 °C, preferably in the range from about 40 °C to about 60 °C, for a period of time ranging from about 0.5 hour to about 20 hours, preferably in the range from about 1 hour to about 15 hours, before completing the
25 reaction at a higher temperature as specified above. Evidently, if the solvent system used is incompatible with the reaction temperature, such as temperatures above 80 °C, then the method may be carried out under pressure.

Examples

30 The following examples is meant to illustrate various embodiments of the invention and should not be read as limiting the scope of protection.

Chromatographic procedures

35 HPLC and GC analyses were performed according to the procedures described below.

Analytical Method HPLC - 5-Chloroindole

Instrument HPLC HP 1100 Agilent	Binary Pump Agilent 1100 Series Detector UV Agilent 1100 Series Column Thermostat Agilent 1100 Series Autosampler Agilent 1100 Series Integration Agilent Chemstation		
Detector	UV 230 nm		
Column	HP Lichrospher C8 250 x 4 mm, 5 μ m		
Column Temperature	40 °C		
Mobile Phase A	Water/Acetonitrile 65:35		
Mobile Phase B	Water/Acetonitrile 15:85		
Flow	1.0 mL/min		
Volume injected	5 μ l		
Run time	45 min		
Gradient	Time	%A	%B
	0	100	0
	30	0	100
	40	0	100
	conditioning		
Runtime	40 min		

Assay against external standard

5

Sample Preparation

Weigh accurately about 50 mg of sample in a 50 mL volumetric flask and add acetonitrile to volume. Transfer 10 mL of obtained solution in a 25 volumetric flask and add acetonitrile to volume. Final concentration 0.2 mg/mL.

10

Standard Preparation

Weigh accurately about 50 mg of Reference Standard in a 50 mL volumetric flask and add acetonitrile to volume. Transfer 10 mL of obtained solution in a 25 volumetric flask and add acetonitrile to volume. Final concentration 0.2 mg/mL

Analytical Procedure

Inject the Standard three times (at least), integrate the obtained chromatograms and calculate Medium Area. If the Standard Deviation % is less than 1.0% inject the Sample and integrate the chromatogram. Calculate the product assay with the formula:

5

$$\text{Assay\%} = (\text{Sample Area} \times \text{Conc. Std} \times 100) / (\text{Standard Area} \times \text{Sample Conc.})$$

Where:

Sample Area = Area obtained by sample injection

10 Standard Area = Average of areas obtained by Standard injection

Sample Conc. = Concentration (mg/ml) of Sample

Standard Conc. = Concentration (mg/ml) of Standard

Analytical Method HPLC - 5-chloro-1-(4-fluorophenyl)-indole

15

Instrument configuration as above except for the gradient.

Mobile Phase A	Water/Acetonitrile 65:35		
Mobile Phase B	Water/Acetonitrile 15:85		
Run time	45 min		
Gradient	Time	%A	%A
	0	60	40
	30	0	100
	40	0	100
	conditioning		
Runtime	40 min		

Assay against external standard

20

Sample Preparation

Weigh accurately about 50 mg of sample in a 50 mL volumetric flask and add acetonitrile to volume. Transfer 10 mL of obtained solution in a 25 volumetric flask and add acetonitrile to volume. Final concentration 0.2 mg/mL.

25

Standard Preparation

Weigh accurately about 50 mg of Reference Standard in a 50 mL volumetric flask and add acetonitrile to volume. Transfer 10 mL of obtained solution in a 25 mL volumetric flask and add acetonitrile to volume. Final concentration 0.2 mg/mL.

5

Analytical Procedure

Inject the Standard three times (at least), integrate the obtained chromatograms and calculate Medium Area. If the Standard Deviation % is less than 1.0% inject the Sample and integrate the chromatogram. Calculate the product assay with the formula:

10

$$\text{Assay\%} = (\text{Sample Area} \times \text{Conc. Std} \times 100) / (\text{Standard Area} \times \text{Sample Conc.})$$

Where:

15

Sample Area = Area obtained by sample injection

Standard Area = Average of areas obtained by Standard injection

Sample Conc. = Concentration (mg/ml) of Sample

Standard Conc. = Concentration (mg/ml) of Standard

Analytical Method GC - 5-chloroindole and 5-chloro-1-(4-fluorophenyl)-indole

Instrument GC	Gc Top 8000 CE Instruments		
Detector	FID		
Column	Zebron (ZB-1)		
	30 m x 0.25 mm		
	0.25 μ m		
Carrier Flow (He)	1.5 mL/min		
Split Flow	50 mL/ml		
H ₂ Flow	30 mL/min		
Air Flow	300 mL/min		
Volume injected	1 μ L		
Run time	25 min		
	Step	Temp (°C)	Duration
	1	120°C	3 min
	1→2	120°→220°C	5 min
	2	220°C	20 min
	Δ T		20 °C/min
T inj	220 °C		
T det	250 °C		

Assay against external standard

5

Internal Standard Solution

Dilute about 2 ml of Undecane (GC Standards) with Acetone in a 250 mL volumetric flask.

Sample Preparation

10

Weigh accurately about 250 mg of sample (5-chloroindole or 5-chloro-1-(4-fluorophenyl)-indole) in a 25 mL volumetric flask and add Internal Standard Solution to volume. Final concentration 25 mg/mL.

Standard Preparation

Weigh accurately about 250 mg of Reference Standard (5-chloroindole or 5-chloro-1-(4-fluorophenyl)-indole) in a 25 mL volumetric flask and add Internal Standard Solution to volume. Final concentration 25 mg/mL.

5

Analytical Procedure

Inject the Standard three times (at least), integrate the obtained chromatograms and calculate the ratio between Area of analyte and Area of Internal Standard. If the ratio Standard Deviation % is less than 1.0% inject the Sample and integrate the chromatogram and calculate the ratio as described above. Calculate the product assay with the formula:

10

$$\text{Assay\%} = (\text{Sample Area Ratio} \times \text{Conc. Std} \times 100) / (\text{Standard Area Ratio} \times \text{Sample Conc.})$$

Where:

15

Sample Area Ratio = Area Ratio obtained by sample injection

Standard Area Ratio = Average of area ratios obtained by Standard injection

Sample Conc. = Concentration (mg/ml) of Sample

Standard Conc. = Concentration (mg/ml) of Standard

20

Analytical Method GC - 5-chloro-1-(4-fluorophenyl)-indole - Conversion In-Process-Control

Instrument configuration as above.

25

Conversion In-Process-Control

Sample Preparation

Stop the stirring and sample 0.1 mL of reaction solution. Dilute with 5 ml of toluene. Filter the solution obtained and inject.

30

Calculate the conversion with the formula:

$$\text{Conversion\%} = (5\text{-chloro-1-(4-fluorophenyl)-indole Area} \times 100) / (5\text{-Chloroindole Area} + 5\text{-chloro-1-(4-fluorophenyl)-indole Area})$$

35

Where:

5-chloro-1-(4-fluorophenyl)-indole Area = Area detected for 5-chloro-1-(4-fluorophenyl)-indole

5 5-Chloroindole Area = Area detected for 5-Chloroindole

Identification of product

NMR spectra were determined on a Bruker Avance 300 spectrometer

10 ^1H -NMR CDCl_3 300MHz (δ ppm, J Hz): 7.70 (1H, d, $J = 2.0$); 7.49-7.39 (3H, m); 7.32 (1H, d, $J = 3.2$); 7.30-7.17 (3H, m); 6.66 (1H, d, $J = 3.2$).

^{13}C -NMR CDCl_3 75MHz (δ ppm, $J_{\text{C,F}}$ Hz): 161.68 (d, $J_{\text{C,F}} = 245.0$); 135.87 (d, $J_{\text{C,F}} = 2.0$); 134.96; 130.62; 129.75; 126.59 (d, $J_{\text{C,F}} = 8.3$); 126.49; 123.18; 120.97; 117.04
15 (d, $J_{\text{C,F}} = 22.0$); 111.71; 103.59.

^{19}F -NMR CDCl_3 282MHz (δ ppm): 114.94 (m).

These data are in agreement with the structure of 5-chloro-1-(4-fluorophenyl)-indole.

20

Synthetic examples with toluene as solvent

Example 1: N,N,N,N-tetramethyl ethylenediamine as ligand

A jacketed glass reactor was charged with 40 g of crude 5-chloro-indole (80% pure as determined by HPLC) (32 g, 0.211 mol), K_2CO_3 (40.2 g, 0.2902 mol), 4-fluoro-
25 bromobenzene (92.3 g, 0.5277 mol), CuI (2.5 g, $1.32 \cdot 10^{-2}$ mol), N,N,N,N-tetramethyl ethylenediamine (3.2 g, $5.28 \cdot 10^{-2}$ mol) and 80 mL of toluene. The mixture was heated to reflux (about 115 °C), under vigorous stirring, and maintained for 40 hours.

After cooling to 60 °C, 80 mL of Toluene and 80 mL of water were added and the mixture was maintained under stirring at 50 °C for ½ hour and the organic layer was
30 separated and treated with 80 mL of water. The residual carbonates were then dissolved by slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was maintained under stirring at 50 °C for ½ hour the aqueous layers were eliminated. The organic layer was then concentrated, by solvent distillation at reduced pressure, and the crude product was obtained as an oil (47.2 g). The yield, based on HPLC (assay against ext. Std.), was about
35 42%.

Example 2: N,N-diethyl ethylenediamine as ligand

Following the procedure of example 1 except that N,N-diethyl ethylenediamine was used in stead of N,N,N,N-tetramethyl ethylenediamine the crude product was obtained as an oil (84 g). The yield, based on HPLC (assay against ext. Std.), was about 50%.

Example 3: Trans-1,2-cyclohexanediamine as ligand

A jacketed glass reactor was charged with 10 g of crude 5-chloro-indole (80% pure as determined by HPLC) (8 g, $5.2 \cdot 10^{-2}$ mol), K_2CO_3 (12.7 g, $9.2 \cdot 10^{-2}$ mol), 4-fluoro-bromobenzene (12.7 g, $7.3 \cdot 10^{-2}$ mol), CuI (1.26 g, $6.6 \cdot 10^{-3}$ mol), *trans*-1,2-cyclohexanediamine (1.13 g, $9.9 \cdot 10^{-3}$ mol) and 20 mL of toluene. The mixture was heated to reflux (about 115 °C), under vigorous stirring, and maintained for 12 hours. The conversion checked by GC was about 79%.

After cooling to 60 °C, the solid residual were filtered off and the organic solution was then concentrated, by solvent distillation at reduced pressure, and the crude product was obtained as an oil (15.4 g)

Example 4: K_3PO_4 as base

A jacketed glass reactor was charged with 20 g of crude 5-chloro-indole (80% pure as determined by HPLC) (16 g, 0.106 mol), K_3PO_4 (18.6 g, 0.088 mol), 4-fluoro-bromobenzene (46.2 g, 0.263 mol), CuI (1.25 g, $1.32 \cdot 10^{-2}$ mol), ethylenediamine (1.58 g, $2.62 \cdot 10^{-2}$ mol) and 40 mL of toluene. The mixture was heated to reflux (about 115 °C), under vigorous stirring, and maintained for 22 hours. An additional amount of K_3PO_4 (9.3 g, $4.4 \cdot 10^{-2}$ mol) was added and the mixture was stirred for 19h. The conversion checked by GC was about 42%.

After cooling to 60 °C, 80 mL of Toluene and 80 mL of water were added and the mixture was maintained under stirring at 50 °C for ½ hour and the organic layer were separated and treated with 80 mL of water. The residual phosphates were then dissolved by slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was maintained under stirring at 50 °C for ½ hour the aqueous layers were eliminated. The organic layer was then concentrated, by solvent distillation at reduced pressure, and the crude product was obtained as an oil (62.3 g).

Example 5: CuBr as catalyst source

A jacketed glass reactor was charged with 40 g of crude 5-chloro-indole (80% pure as determined by HPLC) (32 g, 0.211 mol), K_2CO_3 (40.2 g, 0.2902 mol), 4-fluoro-bromobenzene (92.3 g, 0.5277 mol), CuBr (1.89 g, $1.32 \cdot 10^{-2}$ mol), ethylenediamine (3.2 g,

5.28·10⁻² mol) and 80 ml of toluene. The mixture was heated to reflux (about 115 °C), under vigorous stirring, and maintained for 32 hours. The conversion checked by GC was about 92%.

After cooling to 60 °C, 80 mL of toluene and 80 mL of water were added and the mixture was maintained under stirring at 50 °C for ½ hour and the organic layer was separated and treated with 80 mL of water. The residual carbonates were then dissolved by slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was maintained under stirring at 50 °C for ½ hour the aqueous layers were eliminated. The organic layer was then concentrated, by solvent distillation at reduced pressure, and the crude product was obtained as an oil (64.4 g).

Example 6: CuCl as catalyst source

A jacketed glass reactor was charged with 40 g of crude 5-chloro-indole (80% pure as determined by HPLC) (32 g, 0.211 mol), K₂CO₃ (40.2 g, 0.2902 mol), 4-fluoro-bromobenzene (92.3 g, 0.5277 mol), CuCl (1.31 g, 1.32·10⁻² mol), ethylenediamine (3.2 g, 5.28·10⁻² mol, 25%) and 80 mL of toluene. The mixture was heated to reflux (about 115 °C), under vigorous stirring, and maintained for 32 hours. The conversion checked by GC was about 92%.

After cooling to 60 °C, 80 mL of toluene and 80 mL of water were added and the mixture was maintained under stirring at 50 °C for ½ hour and the organic layer was separated and treated with 80 mL of water. The residual carbonates were then dissolved by slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was maintained under stirring at 50°C for ½ hour the aqueous layers were eliminated. The organic layer was then concentrated, by solvent distillation at reduced pressure, and the crude product was obtained as an oil (7.81g).

Example 7: CuBr₂ as catalyst source

A glass jacketed reactor was charged with 20 g of crude 5-chloro-indole (80% pure as determined by HPLC) (16 g, 0.106 mol), K₂CO₃ (20 g, 0.144 mol), 4-fluoro-bromobenzene (46.1 g, 0.26 mol), CuBr₂ (1.46 g, 6.6·10⁻³ mol), ethylenediamine (1.58 g, 2.6·10⁻² mol) and 40 ml of toluene. The mixture was heated to reflux (about 115 °C), under vigorous stirring, and maintained for 28 hours. The conversion checked by GC was about 44% (after 20 hours the conversion checked by GC was about 43%).

After cooling to 60 °C, 50 mL of Toluene and 40 mL of water were added and the mixture was cooled to 50 °C under stirring. The residual carbonate were then dissolved by slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was

maintained under stirring at 50 °C for ½ hour before the organic layer was separated. The organic layer was treated several times with saturated solution of Sodium Chloride and water under stirring at 50 °C and concentrated, by solvent distillation at reduced pressure. The crude product was obtained as an oil (41 g)

5 Examples 8-18 illustrate variations of the CuI-Ethylenediamine-K₂CO₃-toluene system. They were performed according to the procedure of example 1 except for the details specified. The amounts are given relative to the amount of 5-chloro-indole (calculated as pure 5-chloro-indole). % means mol %, equivalent means molar equivalent, and volume means ml of solvent per g of 5-chloro-indole.

10 *Example 8:*

10% of CuI, 15% of ethylenediamine, 2.1 equivalent of K₂CO₃, 1.1 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 16h reflux. The conversion checked by GC was about 99.5%.

15 *Example 9:*

1% of CuI, 5% of ethylenediamine, 1.5 equivalent of K₂CO₃, 1.1 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 10h reflux. The conversion checked by GC was about 52%.

20 *Example 10:*

1% of CuI, 5% of ethylenediamine, 1.5 equivalent of K₂CO₃, 1.3 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 10h reflux. The conversion checked by GC was about 45%.

25 *Example 11:*

5% of CuI, 15% of ethylenediamine, 1.05 equivalent of K₂CO₃, 1.2 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 18h distilling off water as azeotrope and recycling toluene. The conversion checked by GC was about 55%.

30 *Example 12:*

5% of CuI, 15% of ethylenediamine, 2.1 equivalent of K₂CO₃, 1.1 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 36h reflux. The conversion checked by GC was about 96%.

Example 13:

5% of CuI, 15% of ethylenediamine, 1.5 equivalent of K_2CO_3 , 1.1 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 36h reflux. The conversion checked by GC was about 95%.

5

Example 14:

5% of CuI, 20% of ethylenediamine, 1.1 equivalent of K_2CO_3 , 1.1 equivalent of 4-fluoro-bromobenzene, 2 volumes of Toluene, 44h reflux. The conversion checked by GC was about 99%.

10

Example 15:

5% of CuI, 20% of ethylenediamine, 1.1 equivalent of K_2CO_3 , 2 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 36h reflux. Addition of CuI in two portions (2x2.5%, 2nd after 10 h refluxing). The conversion checked by GC was about 98%.

15

Example 16:

5% of CuI, 1.14 equivalent of ethylenediamine, 1.1 equivalent of K_2CO_3 , 2 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 24h reflux. The conversion checked by GC was about 86%.

20

Example 17:

2.5% of CuI, 40% of ethylenediamine, 1.1 equivalent of K_2CO_3 , 2 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene, 26h reflux. The conversion checked by GC was about 87%.

25

Example 18: Under moderate pressure

5% of CuI, 20% of ethylenediamine, 1.1 equivalent of K_2CO_3 , 2 equivalent of 4-fluoro-bromobenzene, 2 volumes of toluene. The reaction mixture was heated to 120 °C in a closed reactor for 44 h allowing the pressure to increase to a maximum of 2 bar. The conversion checked by GC was about 87%.

30

Toluene and water as solvent system*Example 19: K_3PO_4 as base*

A jacketed glass reactor was charged with 40 g of crude 5-chloro-indole (80% pure as determined by HPLC) (32 g, 0.211 mol), K_3PO_4 (56 g, 0.264 mol), 4-fluoro-bromobenzene

35

(92.3 g, 0.5277 mol), CuI (2.5 g, $1.32 \cdot 10^{-2}$ mol), ethylenediamine (3.2 g, $5.28 \cdot 10^{-2}$ mol), 80 mL of toluene and 20 ml of water. The mixture was heated to reflux (about 115 °C), under vigorous stirring, and maintained for 40 hours. The conversion checked by GC was about 89%.

5 After cooling to 60 °C, 80 mL of Toluene and 80 mL of water were added and the mixture was maintained under stirring at 50 °C for ½ hour and the organic layer was separated and treated with 80 mL of water. The residual phosphates were then dissolved by slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was maintained under stirring at 50 °C for ½ hour the aqueous layers were eliminated. The organic
10 layer was then concentrated, by solvent distillation at reduced pressure, and the crude product was obtained as an oil (86.4 g).

Example 20: K₂CO₃ as base

15 A jacketed glass reactor was charged with 40 g of crude 5-chloro-indole (80% pure as determined by HPLC) (32 g, 0.211 mol), K₂CO₃ (40.2 g, 0.290 mol), 4-fluoro-bromobenzene (92.3 g, 0.5277 mol), CuI (2.5 g, $1.32 \cdot 10^{-2}$ mol), ethylenediamine (3.2 g, $5.28 \cdot 10^{-2}$ mol), 80 ml of toluene and 20 mL of water. The mixture was heated to reflux (about 110 °C), under vigorous stirring, and maintained for 36 hours. The conversion checked by GC was about 67%.

20 After cooling to 60 °C, 80 mL of toluene and 80 mL of water were added and the mixture was maintained under stirring at 50 °C for ½ hour and the organic layer were separated and treated with 80 mL of water. The residual carbonates were then dissolved by slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was maintained under stirring at 50 °C for ½ hour the aqueous layers were eliminated. The organic
25 layer was then concentrated, by solvent distillation at reduced pressure, and the crude product was obtained as an oil (68 g). The yield, based on HPLC (assay against ext. Std.), was about 50%.

Dimethylformamide (DMF) as a solvent

30 *Example 21:*

 A glass jacketed reactor was charged, under nitrogen, with distilled 5-chloro-indole (94% pure as determined by HPLC) (200 g, 1.32 mol), K₂CO₃ (200 g, 1.45 mol), 4-fluoro-bromobenzene (461 g, 2.63 mol), CuI (12.6 g, 0.066 mol), ethylenediamine (15.9 g, 0.26 mol) and 400 mL of dimethylformamide. The mixture was heated to 40°C under vigorous stirring
35 and kept at that temperature for 12 hours whereafter the mixture was to reflux (about 130-

135°C), under vigorous stirring, by increasing the jacket temperature over period of 45 minutes to 145 °C and maintained at reflux for 5 hours.

After cooling to 60 °C, 400 mL of toluene and 400 mL of water were added and the mixture was cooled to 50 °C under stirring. The organic phase was separated and washed, at 50 °C with diluted hydrochloric acid (5 ml HCl 32% + 100 ml H₂O) and finally with a solution of diluted ammonia (5 mL of NH₃ 33% + 200 mL of H₂O). The solvent was then removed by distillation at reduced pressure and the crude product was obtained as an oil (469 g). The yield, based on HPLC (assay against ext. Std.), was about 94%.

Example 22: CuBr as catalyst source

A glass jacketed reactor was charged with 20 g of crude 5-chloro-indole (80% pure as determined by HPLC) (16 g, 0.106 mol), K₂CO₃ (20 g, 0.144 mol), 4-fluoro-bromobenzene (47.7 g, 0.27 mol), CuBr (0.95 g, 6.6*10⁻³ mol), ethylenediamine (1.58 g, 2.6*10⁻² mol) and 40 mL of dimethylformamide. The mixture was heated to reflux (about 130-135 °C), under vigorous stirring, and maintained for 20 hours. The conversion checked by GC was about 99.5% (after 6 hours the conversion checked by GC was about 81%).

After cooling to 60 °C, 80 mL of Toluene and 40 mL of water were added and the mixture was cooled to 50 °C under stirring. The residual carbonate were then dissolved by slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was maintained under stirring at 50 °C for ½ hour. The organic layer was separated and treated with 40 mL of water. The mixture was maintained under stirring at 50 °C for ½ hour the aqueous layers were eliminated. The organic layer was treated several times with saturated solution of ammonium sulphate and water under stirring at 50 °C and then concentrated by solvent distillation at reduced pressure. The crude product was obtained as an oil (38.4 g). The yield, based on HPLC (assay against ext. Std.), was about 80%.

Example 23: CuCl and KI as catalyst source

A glass jacketed reactor was charged with 20 g of crude 5-chloro-indole (80% pure as determined by HPLC) (16 g, 0.106 mol), K₂CO₃ (20 g, 0.144 mol), 4-fluoro-bromobenzene (47.7 g, 0.27 mol), CuCl (0.595 g, 6.0*10⁻³ mol), ethylenediamine (1.58 g, 2.6*10⁻² mol) and 40 mL of dimethylformamide. The mixture was heated to reflux (about 130-135 °C), under vigorous stirring. After 4 hours was added KI (1.16 g, 6.99*10⁻³ mol). The mixture was then maintained at reflux for 16 h. The conversion checked by GC was about 99.5% (after 6 hours the conversion checked by GC was about 53%).

After cooling to 60 °C, 80 mL of Toluene and 40 mL of water were added and the mixture was cooled to 50 °C under stirring. The residual carbonates were then dissolved by

slow addition of aqueous HCl 32% until solution reached pH = 2-3. The mixture was maintained under stirring at 50 °C for ½ hour the organic layer were separated and treated with 40 mL of water. The mixture was maintained under stirring at 50 °C for ½ hour the aqueous layers were eliminated. The organic layer was treated several times with saturated solution of ammonium sulfate and water under stirring at 50 °C then and concentrated, by solvent distillation at reduced pressure. The crude product was obtained as an oil (37.5 g). The yield, based on HPLC (assay against ext. Std.), was about 82%.

Examples 24-29 illustrate variations of the CuI-Ethylenediamine-K₂CO₃-Dimethylformamide system. They were performed according to the procedure of example 21 except for the scale which was 40 g of 5-chloro-indole and the details specified. The amounts are given relative to the amount of 5-chloro-indole (calculated as pure 5-chloro-indole). % means mol %, equivalent means molar equivalent, and volume means ml of solvent per g of 5-chloro-indole.

Example 24

5% of CuI, 20% of ethylenediamine, 1.1 mol of K₂CO₃, 2 mol of 4-fluorobromobenzene, 2 volumes of dimethylformamide, 29h 120°C. The conversion checked by GC was about 80%.

Example 25

5% of CuI, 20% of ethylenediamine, 1.1 mol of K₂CO₃, 2 mol of 4-fluorobromobenzene, 2 volumes of dimethylformamide, 6h 135°C. The conversion checked by GC was about 99%.

Example 26

5% of CuI, 20% of ethylenediamine, 1.1 mol of K₂CO₃, 1.2 mol of 4-fluorobromobenzene, 2 volumes of dimethylformamide. Pretreatment of catalytic system 1 h at 50°C. Reaction 5.5h 135°C. The conversion checked by GC was about 94%.

Example 27

5% of CuI, 20% of ethylenediamine, 1.1 mol of K₂CO₃, 2 mol of 4-fluorobromobenzene, 2 volumes of dimethylformamide and 0.5 volumes of water. Pretreatment of catalytic system 1 h at 50 °C. Reaction 19h 118 °C (reflux). The conversion checked by GC was about 58%.

Example 28

5% of CuI, 20% of ethylenediamine, 1.1 mol of K_2CO_3 , 2 mol of 4-fluorobromobenzene, 2 volumes of Dimethylformamide. Pretreatment of catalytic system 14 h at 50 °C. Reaction 7h 135 °C. The conversion checked by GC was about 92.2%.

Example 29

5% of CuI, 20% of ethylenediamine, 1.1 mol of K_2CO_3 , 2 mol of 4-fluorobromobenzene, 2 volumes of dimethylformamide. NO Pretreatment of catalytic system 50 °C. Reaction 7h 135 °C. The conversion checked by GC was about 78%.

Example 30 illustrates the removal of the impurity 5-bromo-1-(4-fluorophenyl)-indole, which is generated in levels up to 1% by performing a halogen exchange during work-up. Lowering of the impurity by recrystallisation turned out to be difficult.

Example 30

A glass jacketed reactor was charged, under nitrogen, with 5-chloroindole (200 g, 1.32 mol), K_2CO_3 (200 g, 1.45 mol), 4-bromo-fluorobenzene (347 g, 1.98 mol) and 400 ml of dimethylformamide. The mixture was heated to 50°C and ethylenediamine (16 g, 0.26 mol) and CuI (12.5 g, 0.066 mol) were charged in the reactor. The mixture was kept at that temperature for 1.5 hours, then was heated up to 130°C for 1 hour and finally was heated to reflux temperature (about 139°C) for 4 hours. The conversion checked by HPLC was >95%. When the coupling reaction was completed (ref. Example.doc), the mixture was cooled to 100°C and 800ml of toluene were added. After cooling to 60°C the mixture was washed with a solution of diluted ammonia (80 ml of NH_3 30% + 400 ml of H_2O). The organic phase was washed at 40°C with diluted hydrochloric acid (50 ml of HCl 32% + 200 ml of H_2O) and finally with diluted ammonia (44 ml of NH_3 30% + 300 ml of water). The organic solution was concentrated by distillation at normal pressure and then at reduced pressure by stripping with 1-methyl-2-pyrrolidinone (NMP). The residue was diluted with NMP. CuCl (17-35 g, 0.17-0.35 mol) and CuI (2.5g, 0.013 mol) were charged in the reactor and mixture was heated up to 140°C for 6 hours. After diluting with toluene (600 ml), the mixture was filtered and then washed with ammonia (45 ml of NH_3 30% + 300 ml of H_2O). The organic phase was concentrated by distillation at normal pressure, then was diluted with sulfolane and concentrated under vacuum. The crude was finally purified by thin film distillation.

Dioxane as solvent*Example 31: Trans-1,2-Cyclohexanediamine as ligand*

A jacketed glass reactor was charged with 5g of crude 5-chloro-indole (80% pure as determined by HPLC) (4 g, $2.6 \cdot 10^{-2}$ mol), K_2CO_3 (9.58 g, $6.9 \cdot 10^{-2}$ mol), 4-fluorobromobenzene (6.34 g, $3.6 \cdot 10^{-2}$ mol), CuI (0.063 g, $6.6 \cdot 10^{-4}$ mol), *trans*-1,2-cyclohexanediamine (0.377 g, $3.3 \cdot 10^{-3}$ mol) and 33 mL of dioxane. The mixture was heated to about 110 °C, under vigorous stirring, and maintained for 25 hours. The conversion checked by GC was about 45%.

After cooling to 60 °C, the solid residual were filtered off and the organic solution was then concentrated, by solvent distillation at reduced pressure, and the crude product was obtained as an oil (8.2 g).

Neat - Without Solvent*Example 32:*

A jacketed glass reactor was charged with 30 g of distilled 5-Cl-indole (96% pure as determined by HPLC) (28.8 g, 0.190 moles), K_2CO_3 (30.1 g, 0.218 moles), 4-fluorobromobenzene (143.4 g, 0.819 moles), CuI (1.88 g, $9.89 \cdot 10^{-3}$ moles) and ethylenediamine (2.38 g, $3.96 \cdot 10^{-2}$ moles). The mixture was heated to 130-135 °C under vigorous stirring, and maintained for 5 hours.

After cooling to 50 °C, 80 mL of Toluene and 80 mL of water were added and the mixture was maintained under stirring at 50 °C for 15 minutes. The residual carbonates were then dissolved by slow addition of H_2SO_4 36% until solution reached pH = 2-3 (about 40 mL). The mixture was maintained under stirring at 50 °C for ½ hour then cooled to room temperature and stirred overnight. The aqueous layer (upper phase) was eliminated. The organic phase was washed two times with water (2x50mL) and then concentrated, by solvent distillation at reduced pressure. The crude product was obtained as an oil (115.9 g). The yield, based on HPLC (assay against ext. Std.), was about 42%.

Claims

1. Method for manufacture of sertindole comprising manufacturing 5-chloro-1-(4-fluorophenyl)-indole and converting it to sertindole characterised in that the method
5 for manufacture of 5-chloro-1-(4-fluorophenyl)-indole comprises reacting 5-chloro-indole with a 4-fluorophenylhalide in the presence of a base, a chelating ligand and catalytic amounts of a copper salt comprising copper(I) or copper(II) and an anion which does not interfere in an unfavourable way with the reaction.
- 10 2. Method for manufacture of 5-chloro-1-(4-fluorophenyl)-indole comprising reacting 5-chloro-indole with a 4-fluorophenylhalide in the presence of a base, a chelating ligand and catalytic amounts of a copper salt comprising copper(I) or copper(II) and an anion which does not interfere in an unfavourable way with the reaction.
- 15 3. Method according to claim 1 or 2 characterised in that the chelating ligand is a substituted or unsubstituted 1,10-phenanthroline or a compound of the formula $X-(CR^1R^2-(CR^5R^6)_n-CR^3R^4-Y)_m$, wherein X and Y independently are selected from NR^7R^8 and OR^9 , R^1 - R^9 independently are selected from hydrogen, C_{1-6} -alkyl, C_{1-6} -alkyl carboxylic acid and aryl or one of R^1 and R^2 together with one of R^5 and R^6 are C_{3-6} -alkylene, m is 1 or 2 and n is
20 0, 1, 2 or 3.
4. Method according to claim 3 characterised in that the chelating ligand is selected from the group comprising 1,2-cyclohexanediamine, N,N,N,N-tetramethyl ethylenediamine, N,N-diethyl ethylenediamine, ethylenediamine, ethylenediamine N,N,N,N-tetraacetic acid
25 (EDTA), diethylenetriamine N,N,N,N,N-pentaacetic acid (DTPA) and substituted or unsubstituted 1,10-phenanthroline, typically the chelating ligand is selected from 1,2-cyclohexanediamine, N,N,N,N-tetramethyl ethylenediamine, N,N-diethyl ethylenediamine and ethylenediamine, in particular the chelating ligand is ethylenediamine.
- 30 5. Method according to any one of claims 1-4 characterised in that the copper salt is selected from CuCl, CuBr, CuI, CuCl₂, CuBr₂, CuI₂, CuOCOCH₃, Cu(OCOCH₃)₂, anhydrous or hydrated CuSO₄, CuCO₃, Cu₂O and mixtures of said copper salts, typically the copper salt is selected from CuCl, CuBr, CuI, CuCl₂, CuBr₂ or CuI₂.

6. Method according to any one of claims 1-5 characterised in that the 4-fluorophenylhalide is selected from 4-fluoro-bromobenzene or 4-fluoro-iodobenzene, such as 4-fluoro-bromobenzene.
- 5 7. Method according to any one of claims 1-6 characterised in that the 4-fluorophenylhalide is added in a molar surplus relative to the 5-chloro-indole.
8. Method according to claim 7 characterised in that the molar surplus is in the range from 1.1 to 3.
- 10 9. Method according to any one of claims 1-8 characterised in that the catalytic amounts of the copper salt is less than 20 mol % relative to the 5-chloro-indole, typically less than 10 mol % relative to the 5-chloro-indole, such as in the range from about 1 to about 5 mol %.
- 15 10. Method according to any one of claims 1-9 characterised in that the base is selected from the carbonates, hydrogen carbonates, phosphates, hydrogen phosphates, dihydrogen phosphates, oxides and hydroxides of alkali metals.
- 20 11. Method according to claim 10 characterised in that the base is present in a molar excess relative to the 5-chloro-indole, typically the amount of base is in the range from about 1.05 molar equivalents to about 2.5 molar equivalents.
- 25 12. Method according to any one of claims 1-11 characterised in that reaction is completed at temperatures in the range from above 80 °C to 200 °C, typically in the range from 100 °C to 160 °C.

1
INTERNATIONAL SEARCH REPORT

International Application No.

PCT/DK 05/00208

A. CLASSIFICATION OF SUBJECT MATTER

IPC7: C07D 401/14, C07D 209/08

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC7: C07D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

SE,DK,FI,NO classes as above

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

CHEM. ABS DATA, WPI DATA, EPO-INTERNAL

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	J. Am. Chem. Soc., Volume 123, 2001, Artis Klapars et al: "A General and Efficient Copper Catalyst for the Amidation of Aryl Halides and the N-Arylation of Nitrogen Heterocycles", page 7727 - 7729 --	1-12
X	Synlett, Volume 3, 2002, Suk-Ku Kang et al: "Copper-catalyzed N-Arylation of Aryl Iodides with Benzamides or Nitrogen Heterocycles in the Presence of Ethylenediamine", page 427 - page 430 --	1-12
A	US 4710500 A (JENS K. PERREGAARD), 1 December 1987 (01.12.87) --	1-12

☒ Further documents are listed in the continuation of Box C. ☒ See patent family annex.

* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

23 June 2003

Date of mailing of the international search report

25 -06- 2003

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INTERNATIONAL SEARCH REPORT

International Application No.

PCT/DK 92/00208

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	<p>J. Med. Chem., Volume 35, 1992, Jens Perregaard et al: "Noncataleptogenic, Centrally Acting Dopamine D-2 and Serotonin 5-HT2 Antagonists within a Series of 3-Substituted 1-(4-Fluorophenyl)-1H-indoles", page 1092 - page 1101</p> <p style="text-align: center;">-- -----</p>	1-12

INTERNATIONAL SEARCH REPORT

International application No.
PCT/D/00208**Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)**

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:
2. ☒ Claims Nos.: **1-2, 5-12**
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
see next sheet
3. ☐ Claims Nos.:
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.

The expression "chelating agent" in present claims 1-2 defines the compounds by reference to a desirable characteristic. The claims cover all compounds having this characteristic, whereas the application provides support within the meaning of Article 6 PCT and disclosure within the meaning of Article 5 PCT for only a very limited number of such compounds. In the present case, the claims so lack support, and the application so lacks disclosure, that a meaningful search over the whole of the claimed scope is impossible. Independent of the above reasoning, the claims also lack clarity (Article 6 PCT). An attempt is made to define the compounds by reference to a result to be achieved. Again, this lack of clarity in the present case is such as to render a meaningful search over the whole of the claimed scope impossible.

Consequently, the search has been carried out for those parts of the claims which appear to be clear, supported and disclosed, namely those parts relating to the "chelating agents" defined in claims 3-4.

Further the expression "an anion which does not interfere in an unfavourable way with the reaction" is not clear and fully supported by the description. This has however not influenced the search.

INTERNATIONAL SEARCH REPORT

Information on patent family members

02/06/03

International Application No.

PCT/DK 03/00208

Patent document cited in search report			Publication date	Patent family member(s)	Publication date	
US	4710500	A	01/12/87	AU	583607 B	04/05/89
				CA	1256437 A	27/06/89
				CA	1338327 B	14/05/96
				DK	157686 A	11/10/86
				DK	169674 B	09/01/95
				EP	0200322 A,B	05/11/86
				SE	0200322 T3	
				FI	90976 B,C	14/01/94
				FI	861505 A	11/10/86
				HK	116793 A	05/11/93
				IE	58370 B	08/09/93
				IE	860618 L	10/10/86
				JP	1885645 C	22/11/94
				JP	6004587 B	19/01/94
				JP	61236764 A	22/10/86
				LU	90218 A	27/04/98
				US	RE34299 E	29/06/93
				AT	50564 T	15/03/90
				AU	5590186 A	16/10/86
				DE	3669153 D	00/00/00
				DE	19775090 I	00/00/00
				ES	553807 A	01/08/87
				ES	8707499 A	16/10/87
				GB	8509164 D	00/00/00
				GR	860927 A	29/07/86
				IL	78136 A	12/07/90
				LV	5743 A,B	20/10/96
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